



WWJMRD 2022; 8(03): 21-26  
www.wwjmr.com  
International Journal  
Peer Reviewed Journal  
Refereed Journal  
Indexed Journal  
Impact Factor SJIF 2017:  
5.182 2018: 5.51, (ISI) 2020-  
2021: 1.361  
E-ISSN: 2454-6615  
DOI: 10.17605/OSF.IO/TNRCE

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## A Case Study on Solar Distilled Water

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### Abstract

Several developing countries have always had an acute necessity clean, pure drinking water. Sources of water that seem to be brackish (contain chloride ions) and/or contain toxic microbes often aren't suitable for consumption. Additionally, there have been countless coastal areas whereby seawater is copious but fresh drinking water is not. Pure water could also be used in batteries, hospitals, and schools.

Distillation is among the many methods that could be used to treat the water. This demands an usable energy, which can then be delivered by heat or solar irradiance. Water is evaporated in this technique, extracting vapors from colloidal solids, which would then be condensed as distilled water. Desalination process of water from the tap or brackish groundwater can be a delightful, energy-efficient solution for individuals concern about just the quality of them directly by the government provided drinking water and unsatisfied with other methods of supplementary refinement provided to them. This water might well be utilized for a variety of purposes, and one of those is industrial. Distilled water is being used as a feedstock for industrial boilers.

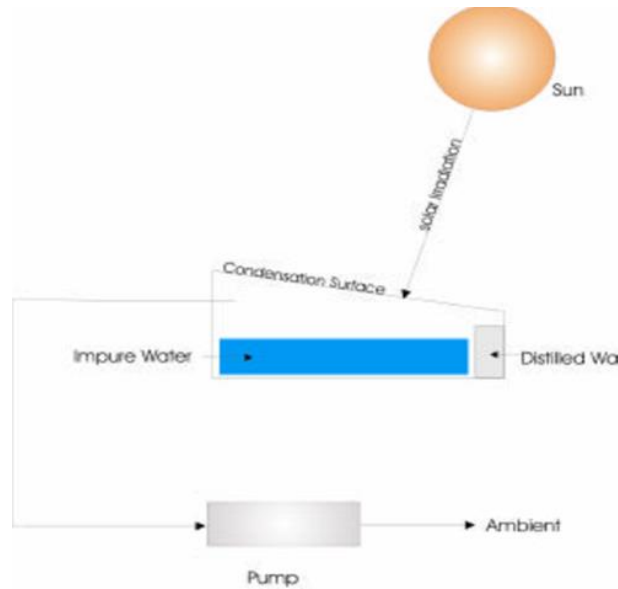
**Keywords:** Desalination, evaporation, boilers, purification, softner, distillation.

### 1. Introduction

Many developing countries have quite an essential need for fresh, pure drinking water. Sources of water that seem to be saline (contain dissolved salts) and/or contain hazardous microbes are often not fit for drinking. Additionally, there are various coastal areas wherever saltwater is plentiful but fresh drinking water is not. Groundwater could also be used in batteries, hospitals, and schools. Distillation along with several other techniques that could be used to purify the water. This demands an usable energy, which can then be delivered by heat or solar irradiance. Water evaporates in this technique, isolating vapors from dissolved salts, which are then condensed as distilled water. Solar water distillation is a solar technique with a lengthy tradition, having sites installed across the globe many years ago. There are indeed a multitude of other approaches for water treatment and desalination, including such solar-powered reverse-osmosis, about which small-scale commercially available technology is accessible. These are not taken into account in this instance. Moreover, if remediation of polluted water is needed but instead of desalination, progressive sand filtration is a feasible alternative.

### 2. Energy requirements and process for water distillation

The heat capacity of the heat of vaporization of freshwater is the required energy to evaporate the water. This one has a kilojoule per kilogram (kJ/kg) rating of 2260. This implies that distillation saltwater takes a heating rate of 2260kJ to generate 1 litre (i.e. 1kg) of distilled water (since the pressure of the liquid is 1kg/litre). This doesn't really allow for the heating technique's accuracy, which would be less than 100%, or any recuperation of temperature difference that is refused when the evaporation is concentrated. This should be emphasized that, although 2260kJ/kg is required to reduce the moisture content, 0.2kJ/kg is necessary to pump a kilogram of moisture through some kind of 20m stream. Distillation is most often contemplated when there is no other neighboring primary source of water that can also be quickly pumped or transported.



A separate still is depicted in Figure 1. The basic operating parameters of all solar stills are about the same. The absorbed sun energy penetrates through into the transparent cover and therefore is generated as heat by a black comes in contact with the filtered water. As a consequence, the fluid is pumped and produces water vapor. The condensation compresses on the transparent cover, which is cooler because it is in contact with the surrounding atmosphere, and trickles down together into drain, where it would be delivered into a storage reservoir.

The here under are the supported in order for an economical solar still for maximum effectiveness, the solar array might still have:

- a high input (undistilled) temperature of the water; and
- a considerable temperature gradient between the water supply and indeed the condensation surface
- minimal vapour leaking A higher stocking temperature of the water can indeed be achieved if:
- a significant portion of incident radiation is captured as heat by the feed solution.

As a response, absorption glass and a strong radiation absorbing surface are required; temperature differences from either the walls and furniture are reduced to a low; and the water is deeper, necessitating less heating. If the precipitating surface absorbed minimal then none of the incoming solar radiation, a large temperature difference can be created. Energy is removed by condensation water, which must be transported promptly from of the collecting surface, for instance, by a subsequent stream of energy or air, whether by concentrating at night.

Single-basin stills have now been thoroughly studied, and their own function is well characterized. Typical performances seem to be in the neighbourhood of 25%. The daily production as a proportion of solar irradiation is maximum in the evening time, because when working fluid is still hot because the outside temperatures is declining. Material selection is very important. Glass or plastic can be used in the cover. Glass should be used for most protracted uses, although a plastic (such as polyethylene) can be employed for relatively brief specific purpose. If indeed the basin of a protracted still would be to be constructed on-site, sands masonry or water-resistant cement is suggested; alternatively, preformed ferro-concrete is an ideal material for large factories stills.

### Multiple-effect basin stills

That has at least one of the following compartments. The uppermost compartment's flooring is the descending compartment's condensation surface. The heat released by the condensing condensate provides the necessary energy to vaporise the feeding water above. Operational productivity therefore is greater than those of a single slope, which would still be normally 35 percent or more, but really the complexity and intricacy are disproportionately higher.

### Wick stills

The feeding water in some kind of a diffuser still steadily streams throughout a porous, radioactive material pad (the wick). There have been two important characteristics to someone using basin stills. To commence, the wick can be slanted so that the feed solution is at a wider picture toward the sun (reducing reflection and presenting a large effective area). Furthermore, so there is less supply water from the still at any particular minute, distilled fluid is pumped more swiftly and to a temperature higher. Simple wick stills seem to be more economical than basin stills, as well as certain configurations are said to have been less inexpensive than a basin have been with the same production.

### Emergency still

Another very simple might still be developed to generate essential drinking water on ground. It wants to take advantage of something like the humanity's dampness. A plastic cover, a bowl or bucket, and a pebble that is all that is essential.

### Output of a solar still

The following algorithm produces an estimation for determining the productivity of a solar distillation.:

$$Q = \frac{E * G * A}{2.3}$$

where:

Q = regular distilled water production (litres/day)

E = for coefficient of performance.

G = the average daily solar illumination (MJ/m<sup>2</sup>).

A = still opening area, i.e. the plan surfaces for something like a basic basin still.

The estimated average worldwide sun illumination in a

representative region is 18.0 MJ/m<sup>2</sup> (5 kWh/m<sup>2</sup>). A simple basin nonetheless seems to have an overall effectiveness with around 30%. As a consequence, the emission per square meter of surface is:

= 2.3 litres (per square metre)

### 3. Methodology

#### Would such a solar panel still serve your expectations?

Individuals require at least or 2 litres of water every day just to thrive. In developing countries, the baseline requirement for normal life (which includes cooking, cleaning, and washing textiles) is 20 gallons per day (in the industrialized world 200 to 400 litres per day is typical). Certain functions, nevertheless, may indeed be completed with seawater, and a basic everyday requirement for filtered water is 5 litres each person. As nothing more than a consequence, 2m<sup>2</sup> of were still necessary so that each person was treated.

Solar stills are generally exclusively sometimes used remove residual salt from sea water. When distinguishing amongst saltwater groundwater and contaminated surface

water, a slow sand filtering and perhaps other purification technology is frequently some less expensive alternative. Even when no fresh moisture is available, the responsible factor are desalination, conveyance, and rainwater storage. Solar stills, as contrasted to seawater desalination technologies, are more appealing as the required output reduces. Stills have always had a capital investment that seems to be approximately commensurate to throughput, but then other technologies benefits from significant economies of scale. As a consequence, again for average household, solar is by far the most cost-effective alternative.

Reverse osmosis or electrodialysis should always be evaluated as a substitute to solar stills for emissions of 1m<sup>3</sup>/day or even above. Much could be driven by the availability and price of energy. For emissions of 200m<sup>3</sup>/day and above and, vapor compression or flash evaporation will normally be the cheapest option. Solar energy systems can help in meeting some of the renewable sources from the latter technologies.

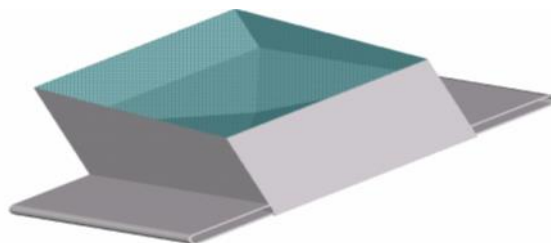


Fig. 2: Isometric View of the Solar Water Distiller.

#### 3.2 Design Specifications:

The Distiller is broken down into the following constituents:

##### 1. Tempered Glass Layer:

Glass really does have the capability of preferentially enabling only high levels of radiation to penetrate through during inhibiting relatively low radiation. This characteristic helps the distilling because it absorbs the preponderance of the outgoing increased energy photons but does not allow it to reflect back. Then it also performs as a condensation surface even though it is always at a cooler temperature than just the water underneath even though it is sensitive to environmental conditions. It is sloped therefore any water molecules which form ultimately activity increase the incline and discharge the condensation into the collection.

##### 2. Primary water reservoir:

Water is separated on the top, immediately behind the glass plate. This fresh water must always be recharged on a frequent basis. The module's floor is painted black to enhance irradiated absorption. The paint shouldn't be soluble in water and therefore must be sun dried. Freshwater is delivered from one location or site to another within different regions of the world through boat, rail, truck, or tube. The involvement in transporting freshwater by car is typical of the same two orders of magnitude as the costs of making water through solar stills. Because of very large quantities, a pipeline may be less expensive. Rainwater gathering is an even simpler technique than thermal desalination in regions where rain is abundant, but it needs a greater space therefore, in most cases, a higher

storage capacity is required. If fully prepared gathering platforms (such like house roofs) are present, these could provide a cheaper priced adequate water supply.

For the purposes of engineering, we presume a very modest power density of around 20%.

Given the highly unpredictable availability of sunlight, which would be greatly dependent on environmental conditions, we should overdesign it for a higher factor of safety — throughout this example.

### 4. Material

#### 4.1 Introduction to Boilers

A boiler seems to be an insulated container that enables ignition heat to be delivered into liquid till it becomes heated water or a gas (steam). The subsequent rise or hot water under compression can thereafter are being used to transport heat to something like a process. Freshwater is an effective and economical medium for delivering heat to something like a process. Whenever water heats up into steam, its handles additional around 1,600 times, delivering a force roughly as destructive as explosives. As a result, steam boiler is exceedingly dangerous equipment dangerous equipment that should be handled with extreme precaution. Boilers was being used in basic forms for the last several centuries, but advancement was postponed due to primitive building techniques and indeed the operation becoming incredibly risky. Nevertheless, even by half 20th century, boilers was becoming the major source of power for manufacturing processes and conveyance. Water as little more than a heat transfer medium provides various advantages. Freshwater is very affordable, easy to handle, and the vapor is invisible, odorless, and of enormously high

quality. Evaporative cooling relates to the procedure of heating a liquid till it becomes gaseous. Heat is then transferred through one body to another through (1) radiation, which would be the heat transfer process from a hot body to a cold body without physical interaction, (2) convection, which is really the heat transfer process by a conveying medium such as air or water, and (3) conduction, which would be the heat transfer process by physical interaction.

**4.2 Boilers type**

There are practically limitless boiler models, but they probably fall into one of two main categories: (1) Fire tube or "fire in tube" boilers, which encompass long steel pipes through which hot pressures from either a heating system finish and upon which the water to really be transformed to vapor propagates, and (2) Watertube or "water in tube" boilers, which encompass the circumstances overturned with both the current moving through into the tubes and indeed the furnace for the hot fumes is manufactured of water. Heat (gases) from the combustion of the fuel move through tubes and are transmitted to waters in a large cylindrical storage area in some kind of a fire tube boiler fire tube boilers.

Fire tube boilers have just a less initial cost, seem to be more fuel efficient, and are easier than running, even though they are normally limited to capacities of 50,000 PPH and pressure of 250 PSIG. Water tube boilers are divided into different types: "D" type, "A" type, "O" type, bent tube, and cast-iron sectional. All fire tube boilers and the overwhelming bulk of water tube boilers are packaged boilers, which may have been carried by truck, rail, or barge. Field constructed boilers are big water tube boilers used in industries with high steam demands and then in utilities. Vertical tubeless boilers are often used for low loads but should not fall into the other category because they lack tubes. Boilers and pressure vessels are manufactured in accordance with the requirements of the American Society of Mechanical Engineers, or ASME, abbreviated as the "ASME Code." High pressure boilers are

fired vessels with pressure greater than 15psig and a temperature greater than 160oF that have been constructed in line with Section I of the ASME Code and exhibit the ASME S stamp.

**4.3.1 Water Softners**

Freshwater, whenever it pours over the earth, along caves and springs, scoops up several of the components from either the limestone as well as some other environmental occurrences that disintegrate and stay. Such components are referred to collectively as hardness. Grandma's tea kettle, used it as an example throughout Chapter One, seemed to have a "build up" in the bottom where she had to clean out on such a constant schedule, usually with vinegar. Such "accumulation" is known to as hardness. In a massive engineering steam power plant, the liquid might well be totally replenished as regularly at once per hour. Presumably, with needs to be considered, temperatures, and pressures than the tea kettle, the boiler would eventually develop scale from any of this hardness, reducing and subsequently preventing water circulation and heat transmission, damaging the boiler. Proper elimination of foreign matter from either the boiler's feed solution became increasingly crucial as the boiler's operating pressure rises. Large utility boilers operating at 3,000 PSIG or above may potentially use filtered water for maximum purification. The primary purpose of a filtration system seems to be to eliminate hardness from boiler treated wastewater. Makeup water is water supplied from of the municipal water system, well water, or perhaps other sources for the increase of the number water to the boiler system to compensate the water dissipated. Substantial filtering of the water might occur in the water softener, but that's not the objective of its creation, as well as many other pollutants in the water could actually taint the filtration system, harming its performance. Hardness is predominantly comprised of calcium (Ca) and magnesium (Mg), with varying proportions of sodium (Na), potassium (P), and various other components.



Fig. 3: Watertube Boiler D-Type



Firetube Scotch Marine

Measurement was performed in grains, including one grain containing 17.1 ppm of these components in water. The justification for just using hardness as the system of measurement would be that quantifying in parts per million (ppm) is far more complicated and expensive. The hardness ranges form location to another. Hardness is typically exceptionally low near saline water because dolomite

seems to be almost non-existent, but hardness is generally fairly significant in mountainous regions where limestone is abundant. All water softeners mellow or lessen hardness from the water. Calcium (Ca++) and Magnesium (Mg++) are the primary minerals in water that produce "hard" water.

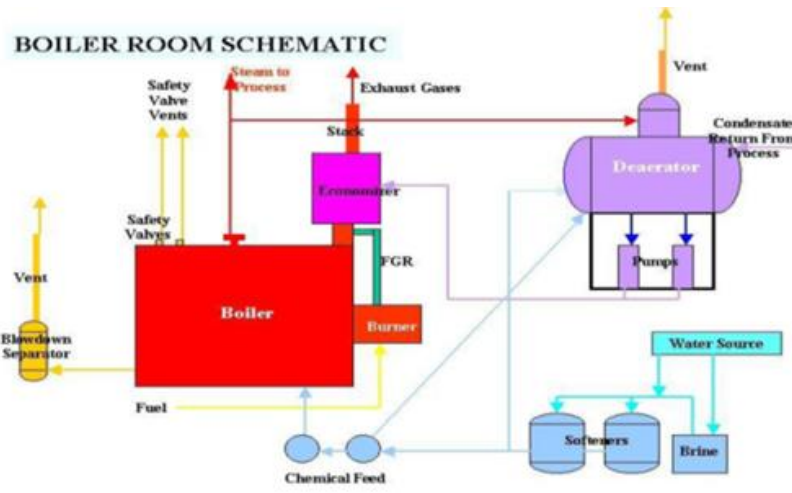


Fig. 4: Boiler room schematic.

Although displacement occurred, this procedure might simply be described to as "ion substitution." Sodium (Na+) ions, which are "soft," are exchanged or traded for Calcium and Magnesium whenever the flow passes through into the softener tank. The softening agent is also referred as resin or Zeolite. Polystyrene resin is the right description for it. The resin seems to have the potential to attract positive charges towards itself. It is doing it because it acquires a negative charge during production. It is a fundamental rule because unlike charges attract, i.e., a negative should attract a positive and behave accordingly. Hundreds or even thousands of Zeolite beads are contained in a softener container. Each bead is intrinsically negative and might even be discharged or regenerated utilizing positive ions. The Zeolite in some kind of a softener is loaded with positive, "soft" sodium ions. Calcium and Magnesium ions become increasingly drawn to the bead as "hard" water flows through into the Zeolite. As the "hard" ions connect to the Zeolite bead, they expel the "soft" Sodium ions who are already there. In consequence, the Sodium there in water system is "interchanged" for the Calcium and Magnesium, with both the Calcium and Magnesium remaining on the Zeolite beads and indeed the Sodium ions assuming their places in the water rushing through to the softener container. Soft water goes out of another container

as a byproduct of this "exchange" procedure. It has become clear that somehow a softener would continue to create "soft" water as just as long as Sodium ions exist on the Zeolite beads to "exchange" with both the Calcium and Magnesium ions mostly in "hard" water. Because when supplies of Sodium ions run out, the Zeolite beads must always be "regenerated" utilizing incoming Sodium ions. A multiple procedure is used to rejuvenate the Zeolite beads.

**4.3.2 Softner design**

Water softeners are produced in solitary mineral container (simplex), double mineral container (duplex), and multiple mineral container designs. Because regeneration rounds typically require up to several hours, simplex machines are only utilized when this interruption can be sustained. Duplicate machines are often used to minimize interruptions by permitting the regenerating among one unit to start happening whereas the second unit is always on line. Triplex or other multiplex units are commonly incorporated because to a necessity for expanded capacity, and units can be implemented to preserve soft water availability. Due to the obvious trustworthiness of sophisticated computing metering/controls for regenerating, customers could now rely upon separate components with some more frequently renewal.



Fig. 5: Simplex Softener

Duplex Softener

Triplex Softener

**4.3.3. Softner sizing formula:**

$$C = \frac{M * H * T}{R}$$

C = Softener capability in cubic feet of resin  
 M = Adjustment water volume per hour in gallons; this

same proportion that must be softened.  
 T = Preferred time in hours before charge and discharge cycles  
 H = Water hardness in particles  
 R = Resin Capacity each Cubic Feet.

Compounds can always be introduced in either house by such a range of methodologies or a combination of techniques. Continuous-flow- flow pumps are the most economical and efficient technique for high-pressure steam installations. Continuous feed pumps provide us with a steady, continuous stream of chemicals with little to no rising or falling swings or residue. The feed pumps must be properly selected and managed accordingly to provide maximum performance and reliability. A compressor is sometimes used to feed treatment chemicals into the steam drum or the boiler feedwater line. Another feed pump is used to pump stainless steel quills to pump amine into in the steam header, whereas a third pump introduces an oxygen scavenger into to the evaporator section.

## 5. Conclusions

The yield of a solar still (distillate) is a crucial characteristic of solar radiation on something like a horizontal plane. For a constant room temperature, the conversion efficiency increases with sun solar irradiance. The depth of water in the basin also has a significant impact on performance. The greatest efficiency of a single basin planetary system is still around 60%. The variety of transparent slipcovers in a solar really does not a strong motivating factor since it raises the temperature of inner cover, producing with much less vapour pressure condensation. We would see that the range of possibilities in desalination is rather broad. Since there will always be a need for freshwater, treatment systems must be developed towards becoming cleaner, better efficient, and more altruistic.

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